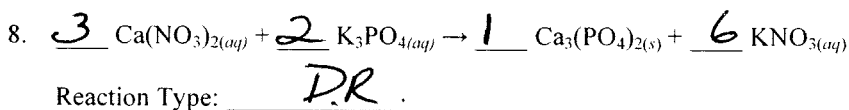
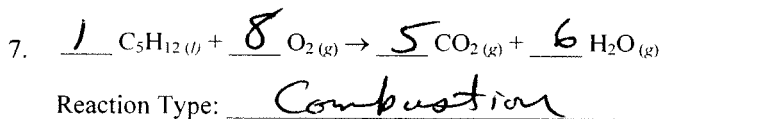
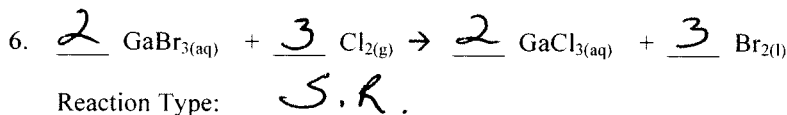
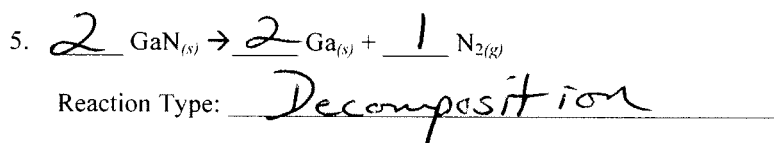
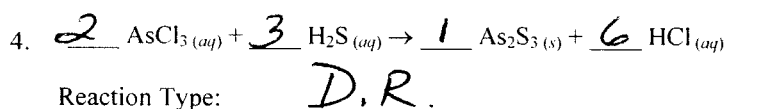
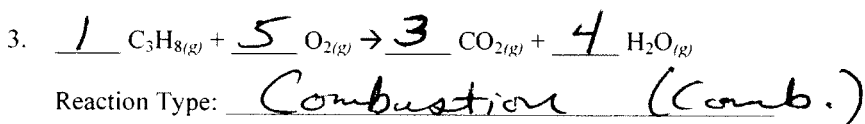
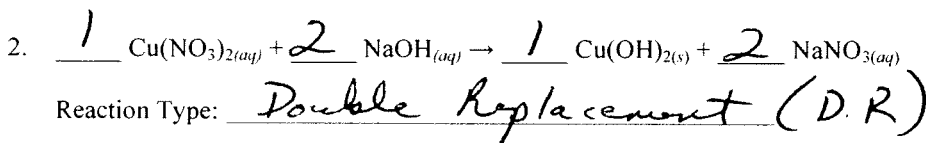
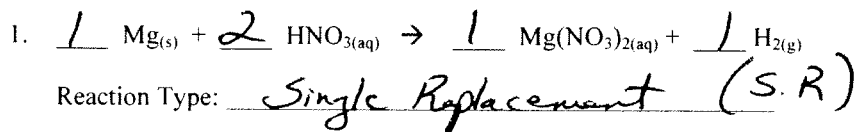


Chemistry 192 • Reactions & Equations 1

CLASSIFY each of the following reactions and BALANCE the equations.

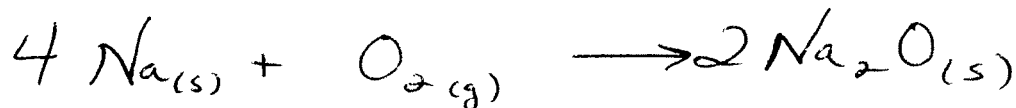
KEY



9. Sodium metal is exposed to oxygen in the air, and forms the white solid, sodium oxide.

Reaction Type: synthesis (combination)

Complete, Balanced Equation:



10. Aqueous sodium phosphate reacts with aqueous cobalt (II) chloride to form the precipitate, cobalt (II) phosphate, with sodium chloride remaining in solution.

Reaction Type: D.R.

Complete, Balanced Equation:

