

Standard Thermodynamic Values for Selected Substances at 298 K

Substance or Ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol·K)	Substance or Ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol·K)	Substance or Ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol·K)
e ⁻ (g)	0	0	20.87	ClO ₂ (g)	102	120	256.7	OH ⁻ (aq)	-229.94	-157.30	-10.54
Aluminum				Cl ₂ O(g)	80.3	97.9	266.1	H ₂ O(g)	-241.826	-228.60	188.72
Al(s)	0	0	28.3	Chromium				H ₂ O(l)	-285.840	-237.192	69.940
Al ³⁺ (aq)	-524.7	-481.2	-313	Cr(s)	0	0	23.8	H ₂ O ₂ (l)	-187.8	-120.4	110
AlCl ₃ (s)	-704.2	-628.9	110.7	Cr ³⁺ (aq)	-1971	—	—	H ₂ O ₂ (aq)	-191.2	-134.1	144
Al ₂ O ₃ (s)	-1676	-1582	50.94	CrO ₄ ²⁻ (aq)	-863.2	-706.3	38	Phosphorus			
Barium				Cr ₂ O ₇ ²⁻ (aq)	-1461.	-1257	214	P ₄ (s, white)	0	0	41.1
Ba(s)	0	0	62.5	Copper				P(g)	314.6	278.3	163.1
Ba(g)	175.6	144.8	170.28	Cu(s)	0	0	33.1	P(s, red)	-17.6	-12.1	22.8
Ba ²⁺ (g)	1649.9	—	—	Cu(g)	341.1	301.4	166.29	P ₂ (g)	144	104	218
Ba ²⁺ (aq)	-538.36	-560.7	13	Cu ⁺ (aq)	51.9	50.2	-26	P ₄ (g)	58.9	24.5	280
BaCl ₂ (s)	-806.06	-810.9	126	Cu ²⁺ (aq)	64.39	64.98	-98.7	PCl ₃ (g)	-287	-268	312
BaCO ₃ (s)	-1219	-1139	112	Cu ₂ O(s)	-168.6	-146.0	93.1	PCl ₃ (l)	-320	-272	217
BaO(s)	-548.1	-520.4	72.07	CuO(s)	-157.3	-130	42.63	PCl ₅ (g)	-402	-323	353
BaSO ₄ (s)	-1465	-1353	132	Cu ₂ S(s)	-79.5	-86.2	120.9	PCl ₅ (s)	-443.5	—	—
Boron				CuS(s)	-53.1	-53.6	66.5	P ₄ O ₁₀ (s)	-2984	-2698	229
B(β-rhombohedral)	0	0	5.87	Fluorine				PO ₄ ³⁻ (aq)	-1266	-1013	-218
BF ₃ (g)	-1137.0	-1120.3	254.0	F ₂ (g)	0	0	202.7	HPO ₄ ²⁻ (aq)	-1281	-1082	-36
BCl ₃ (g)	-403.8	-388.7	290.0	F(g)	78.9	61.8	158.64	H ₂ PO ₄ ⁻ (aq)	-1285	-1135	89.1
B ₂ H ₆ (g)	35	86.6	232.0	F ⁻ (g)	-255.6	-262.5	145.47	H ₃ PO ₄ (aq)	-1277	-1019	228
B ₂ O ₃ (s)	-1272	-1193	53.8	F ⁻ (aq)	-329.1	-276.5	-9.6	Potassium			
H ₃ BO ₃ (s)	-1094.3	-969.01	88.83	HF(g)	-273	-275	173.67	K(s)	0	0	64.672
Bromine				Hydrogen				K(g)	89.2	60.7	160.23
Br ₂ (l)	0	0	152.23	H ₂ (g)	0	0	130.6	K ⁺ (g)	514.197	481.202	154.47
Br ₂ (g)	30.91	3.13	245.38	H(g)	218.0	203.30	114.60	K ⁺ (aq)	-251.2	-282.28	103
Br(g)	111.9	82.40	174.90	H ⁺ (aq)	0	0	0	KF(s)	-568.6	-538.9	66.55
Br ⁻ (g)	-218.9	—	—	H ⁺ (g)	1536.3	1517.1	108.83	KCl(s)	-436.7	-409.2	82.59
Br ⁻ (aq)	-120.9	-102.82	80.71	Iodine				KBr(s)	-394	-380	95.94
HBr(g)	-36.3	-53.5	198.59	I ₂ (s)	0	0	116.14	KI(s)	-328	-323	106.39
Cadmium				I ₂ (g)	62.442	19.38	260.58	KOH(s)	-424.8	-379.1	78.87
Cd(s)	0	0	51.5	I(g)	106.8	70.21	180.67	KClO ₃ (s)	-397.7	-296.3	143.1
Cd(g)	112.8	78.20	167.64	I ⁻ (g)	-194.7	—	—	KClO ₄ (s)	-432.75	-303.2	151.0
Cd ²⁺ (aq)	-72.38	-77.74	-61.1	I ⁻ (aq)	-55.94	-51.67	109.4	Rubidium			
CdS(s)	-144	-141	71	HI(g)	25.9	1.3	206.33	Rb(s)	0	0	69.5
Calcium				Iron				Rb(g)	85.81	55.86	169.99
Ca(s)	0	0	41.6	Fe(s)	0	0	27.3	Rb ⁺ (g)	495.04	—	—
Ca(g)	192.6	158.9	154.78	Fe ³⁺ (aq)	-47.7	-10.5	-293	Rb ⁺ (aq)	-246	-282.2	124
Ca ²⁺ (g)	1934.1	—	—	Fe ²⁺ (aq)	-87.9	-84.94	113	RbF(s)	-549.28	—	—
Ca ²⁺ (aq)	-542.96	-553.04	-55.2	FeCl ₂ (s)	-341.8	-302.3	117.9	RbCl(s)	-435.35	-407.8	95.90
CaF ₂ (s)	-1215	-1162	68.87	FeCl ₃ (s)	-399.5	-334.1	142	RbBr(s)	-389.2	-378.1	108.3
CaCl ₂ (s)	-795.0	-750.2	114	FeO(s)	-272.0	-251.4	60.75	RbI(s)	-328	-326	118.0
CaCO ₃ (s)	-1206.9	-1128.8	92.9	Fe ₂ O ₃ (s)	-825.5	-743.6	87.400	Silicon			
CaO(s)	-635.1	-603.5	38.2	Fe ₃ O ₄ (s)	-1121	-1018	145.3	Si(s)	0	0	18.0
Ca(OH) ₂ (s)	-986.09	-898.56	83.39	Lead				SiF ₄ (g)	-1614.9	-1572.7	282.4
Ca ₃ (PO ₄) ₂ (s)	-4138	-3899	263	Pb(s)	0	0	64.785	SiO ₂ (s)	-910.9	-856.5	41.5
CaSO ₄ (s)	-1432.7	-1320.3	107	Pb ²⁺ (aq)	1.6	-24.3	21	Silver			
Carbon				PbCl ₂ (s)	-359	-314	136	Ag(s)	0	0	42.702
C(graphite)	0	0	5.686	PbO(s)	-218	-198	68.70	Ag(g)	289.2	250.4	172.892
C(diamond)	1.896	2.866	2.439	PbO ₂ (s)	-276.6	-219.0	76.6	Ag ⁺ (aq)	105.9	77.111	73.93
C(g)	715.0	669.6	158.0	PbS(s)	-98.3	-96.7	91.3	AgF(s)	-203	-185	84
				PbSO ₄ (s)	-918.39	-811.24	147	AgCl(s)	-127.03	-109.72	96.11

Substance or Ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol·K)
CO(g)	-110.5	-137.2	197.5
CO ₂ (g)	-393.5	-394.4	213.7
CO ₂ (aq)	-412.9	-386.2	121
CO ₃ ²⁻ (aq)	-676.26	-528.10	-53.1
HCO ₃ ⁻ (aq)	-691.11	587.06	95.0
H ₂ CO ₃ (aq)	-698.7	-623.42	191
CH ₄ (g)	-74.87	-50.81	186.1
C ₂ H ₂ (g)	227	209	200.85
C ₂ H ₄ (g)	52.47	68.36	219.22
C ₂ H ₆ (g)	-84.667	-32.89	229.5
C ₃ H ₈ (g)	-105	-24.5	269.9
C ₄ H ₁₀ (g)	-126	-16.7	310
C ₆ H ₆ (l)	49.0	124.5	172.8
CH ₃ OH(g)	-201.2	-161.9	238
CH ₃ OH(l)	-238.6	-166.2	127
HCHO(g)	-116	-110	219
HCOO ⁻ (aq)	-410	-335	91.6
HCOOH(l)	-409	-346	129.0
HCOOH(aq)	-410	-356	164
C ₂ H ₅ OH(g)	-235.1	-168.6	282.6
C ₂ H ₅ OH(l)	-277.63	-174.8	161
CH ₃ CHO(g)	-166	-133.7	266
CH ₃ COOH(l)	-487.0	-392	160
C ₆ H ₁₂ O ₆ (s)	-1273.3	-910.56	212.1
C ₁₂ H ₂₂ O ₁₁ (s)	-2221.7	-1544.3	360.24
CN ⁻ (aq)	151	166	118
HCN(g)	135	125	201.7
HCN(l)	105	121	112.8
HCN(aq)	105	112	129
CS ₂ (g)	117	66.9	237.79
CS ₂ (l)	87.9	63.6	151.0
CH ₂ Cl ₂ (g)	-83.7	-60.2	234
CH ₂ Cl ₂ (l)	-117	-63.2	179
CHCl ₃ (l)	-132	-71.5	203
CCl ₄ (g)	-96.0	-53.7	309.7
CCl ₄ (l)	-139	-68.6	214.4
COCl ₂ (g)	-220	-206	283.74
Cesium			
Cs(s)	0	0	85.15
Cs(g)	76.7	49.7	175.5
Cs ⁺ (g)	458.5	427.1	169.72
Cs ⁺ (aq)	-248	-282.0	133
CsF(s)	-554.7	-525.4	88
CsCl(s)	-442.8	-414	101.18
CsBr(s)	-395	-383	121
CaI(s)	-337	-333	130
Chlorine			
Cl ₂ (g)	0	0	223.0
Cl(g)	121.0	105.0	165.1
Cl ⁻ (g)	-234	-240	153.25
Cl ⁻ (aq)	-167.46	-131.17	55.10
HCl(g)	-92.31	-95.30	186.79
HCl(aq)	-167.46	-131.17	55.06

Substance or Ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol·K)
Lithium			
Li(s)	0	0	29.10
Li(g)	161	128	138.67
Li ⁺ (g)	687.163	649.989	132.91
Li ⁺ (aq)	-278.46	-293.8	14
LiF(s)	-616.9	-588.7	35.66
LiCl(s)	-408	-384	59.30
LiBr(s)	-351	-342	74.1
LiI(s)	-270	-270	85.8
Magnesium			
Mg(s)	0	0	32.69
Mg(g)	150	115	148.55
Mg ²⁺ (g)	2351	—	—
Mg ²⁺ (aq)	-461.96	-456.01	118
MgCl ₂ (s)	-641.6	-592.1	89.630
MgCO ₃ (s)	-1112	-1028	65.86
MgO(s)	-601.2	-569.0	26.9
Mg ₃ N ₂ (s)	-461	-401	88
Manganese			
Mn(s, α)	0	0	31.8
Mn ²⁺ (aq)	-219	-223	-84
MnO ₂ (s)	-520.9	-466.1	53.1
MnO ₄ ⁻ (aq)	-518.4	-425.1	190
Mercury			
Hg(l)	0	0	76.027
Hg(g)	61.30	31.8	174.87
Hg ²⁺ (aq)	171	164.4	-32
Hg ₂ ²⁺ (aq)	172	153.6	84.5
HgCl ₂ (s)	-230	-184	144
Hg ₂ Cl ₂ (s)	-264.9	-210.66	196
HgO(s)	-90.79	-58.50	70.27
Nitrogen			
N ₂ (g)	0	0	191.5
N(g)	473	456	153.2
N ₂ O(g)	82.05	104.2	219.7
NO(g)	90.29	86.60	210.65
NO ₂ (g)	33.2	51	239.9
N ₂ O ₄ (g)	9.16	97.7	304.3
N ₂ O ₅ (g)	11	118	346
N ₂ O ₅ (s)	-43.1	114	178
NH ₃ (g)	-45.9	-16	193
NH ₃ (aq)	-80.83	26.7	110
N ₂ H ₄ (l)	50.63	149.2	121.2
NO ₃ ⁻ (aq)	-206.57	-110.5	146
HNO ₃ (l)	-173.23	-79.914	155.6
HNO ₃ (aq)	-206.57	-110.5	146
NF ₃ (g)	-125	-83.3	260.6
NOCl(g)	51.71	66.07	261.6
NH ₄ Cl(s)	-314.4	-203.0	94.6
Oxygen			
O ₂ (g)	0	0	205.0
O(g)	249.2	231.7	160.95
O ₃ (g)	143	163	238.82

Substance or Ion	ΔH_f° (kJ/mol)	ΔG_f° (kJ/mol)	S° (J/mol·K)
AgBr(s)	-99.51	-95.939	107.1
AgI(s)	-62.38	-66.32	114
AgNO ₃ (s)	-45.06	19.1	128.2
Ag ₂ S(s)	-31.8	-40.3	146
Sodium			
Na(s)	0	0	51.446
Na(g)	107.76	77.299	153.61
Na ⁺ (g)	609.839	574.877	147.85
Na ⁺ (aq)	-239.66	-261.87	60.2
NaF(s)	-575.4	-545.1	51.21
NaCl(s)	-411.1	-384.0	72.12
NaBr(s)	-361	-349	86.82
NaOH(s)	-425.609	-379.53	64.454
Na ₂ CO ₃ (s)	-1130.8	-1048.1	139
NaHCO ₃ (s)	-947.7	-851.9	102
NaI(s)	-288	-285	98.5
Strontium			
Sr(s)	0	0	54.4
Sr(g)	164	110	164.54
Sr ²⁺ (g)	1784	—	—
Sr ²⁺ (aq)	-545.51	-557.3	-39
SrCl ₂ (s)	-828.4	-781.2	117
SrCO ₃ (s)	-1218	-1138	97.1
SrO(s)	-592.0	-562.4	55.5
SrSO ₄ (s)	-1445	-1334	122
Sulfur			
S(rhombic)	0	0	31.9
S(monoclinic)	0.3	0.096	32.6
S(g)	279	239	168
S ₂ (g)	129	80.1	228.1
S ₈ (g)	101	49.1	430.211
S ²⁻ (aq)	41.8	83.7	22
HS ⁻ (aq)	-17.7	12.6	61.1
H ₂ S(g)	-20.2	-33	205.6
H ₂ S(aq)	-39	-27.4	122
SO ₂ (g)	-296.8	-300.2	248.1
SO ₃ (g)	-396	-371	256.66
SO ₄ ²⁻ (aq)	-907.51	-741.99	17
HSO ₄ ⁻ (aq)	-885.75	-752.87	126.9
H ₂ SO ₄ (l)	-813.989	-690.059	156.90
H ₂ SO ₄ (aq)	-907.51	-741.99	17
Tin			
Sn(white)	0	0	51.5
Sn(gray)	3	4.6	44.8
SnCl ₄ (l)	-545.2	-474.0	259
SnO ₂ (s)	-580.7	-519.7	52.3
Zinc			
Zn(s)	0	0	41.6
Zn(g)	130.5	94.93	160.9
Zn ²⁺ (aq)	-152.4	-147.21	-106.5
ZnO(s)	-348.0	-318.2	43.9
ZnS(s, zinc blende)	-203	-198	57.7