Formulas & Naming 5: General

Write formulas for the following compounds:

1. Magnesium sulfide _____
2. Sulfur trioxide _____
3. Sodium phosphide _____
4. Copper(II) oxide _____
5. Hydrocyanic acid _____
6. Sulfur hexafluoride _____
7. Copper (I) oxide _____
8. Tetraphosphorus decaoxide _____
9. Cesium iodide _____
10. Hypochlorous acid _____
11. Chromium(III) sulfate _____
12. Ammonium bromide _____
13. Phosphoric acid _____
14. Oxygen dichloride _____
15. Ammonium sulfite _____
16. Iron (III) hydroxide _____
17. Hydrochloric acid _____
18. Dinitrogen monoxide _____
19. Chlorous acid _____
20. Dinitrogen pentaoxide _____
21. aluminum dichromate _____
22. sodium hydrogen phosphate _____
23. nickel (II) cyanide _____
24. potassium hydride _____
25. iron (II) acetate _____

Write names for the following compounds:

26. Na_2SO_3 _____
27. $Fe_3(PO_4)_2$ _____
28. $OsCl_4$ _____
29. HBr _____
30. P_2O _____
31. $Ca(CN)_2$ _____
32. OF_2 _____
33. H_2S _____
34. Mn_2O_7 _____
35. $PbCl_4$ _____
36. SeF_4 _____
37. $Re_2(SO_3)_3$ _____
38. $HClO_4$ _____
39. $Sr(NO_3)_2$ _____
40. NO_2 _____
41. KH_2PO_4 _____
42. H_2CO_3 _____
43. $CoCO_3$ _____
44. CS_2 _____
45. AgBr _____
46. NaOH _____
47. $AuCl_3$ _____
48. $In_2(CrO_4)$ _____
49. Na_2Se _____
50. Ra_3N_2 _____