

Math Review & Problem-Solving 5

Unit Conversions • Chemistry 210

KEY

Carry out the following conversions. Express the answers to the correct number of significant figures. (Use the back cover of the book for some conversion factors.)

1) $0.00232 \text{ g} \rightarrow \mu\text{g}$

$$\frac{.00232 \text{ g}}{1 \times 10^{-6} \text{ g}} = 2320$$

2) $5130 \text{ g} \rightarrow \text{kg}$

$$5.13 \text{ kg}$$

3) $200 \text{ cm} \rightarrow \text{dm}$

$$20 \text{ dm}$$

4) $200 \text{ cm}^2 \rightarrow \text{dm}^2$

$$\frac{200 \text{ cm}^2}{100 \text{ cm}^2} = 2 \text{ dm}^2$$

5) $200 \text{ cm}^3 \rightarrow \text{dm}^3$

$$\frac{200 \text{ cm}^3}{1000 \text{ cm}^3} = 0.2 \text{ dm}^3$$

6) $3.5 \times 10^5 \text{ ng} \rightarrow \text{mg}$

$$\frac{3.5 \times 10^5 \text{ ng}}{1 \times 10^9 \text{ ng}} = 0.35 \text{ mg}$$

7) $8.43 \times 10^{12} \text{ mg} \rightarrow \text{Gg}$

$$8.43 \text{ Gg}$$

8) $25 \text{ miles} \rightarrow \text{m}$

$$\frac{25 \text{ miles}}{0.6214 \text{ mi}} \times \frac{1 \text{ km}}{1 \text{ km}} \times \frac{1000 \text{ m}}{1 \text{ km}} = 40. \text{ m}$$

9) $0.0038 \text{ dL} \rightarrow \mu\text{L}$

$$380 \mu\text{L}$$

10) $0.0344 \text{ g/mL} \rightarrow \text{g/L}$

$$\frac{0.0344 \text{ g}}{1 \text{ mL}} \times \frac{1000 \text{ mL}}{1 \text{ L}} = 34.4 \text{ g/L}$$

11) $3500 \text{ kg} \rightarrow \text{mg}$

$$3500.000.000$$

$$3.5 \times 10^9 \text{ mg}$$

12) $3500 \text{ kg} \rightarrow \mu\text{g}$

$$3.5 \times 10^{12} \mu\text{g}$$

13) $3500 \text{ kg} \rightarrow \text{Mg}$

$$3.5 \text{ Mg}$$

14) $1.11 \times 10^7 \text{ ns} \rightarrow \text{ms}$

$$11.1 \text{ ms}$$

15) $785 \text{ mg} \rightarrow \text{dg}$

$$7.85 \text{ dg}$$

16) $0.0088 \text{ mg} \rightarrow \text{pg}$

$$\frac{0.0088 \text{ mg}}{1000 \text{ mg}} \times \frac{1 \times 10^{12} \text{ pg}}{1 \text{ mg}} = 8.8 \times 10^6 \text{ pg}$$

17) $2.5 \times 10^{11} \text{ dg} \rightarrow \text{Gg}$

$$\frac{2.5 \times 10^{11} \text{ dg}}{100 \text{ dg}} \times \frac{1 \text{ g}}{1 \text{ g}} \times \frac{1 \text{ Gg}}{1 \times 10^9 \text{ g}} = 25 \text{ Gg}$$

18) $3.22 \times 10^5 \text{ mL} \rightarrow \text{dm}^3$

$$\frac{3.22 \times 10^5 \text{ mL}}{1000 \text{ mL}} \times \frac{1 \text{ L}}{1 \text{ L}} \times \frac{1 \text{ dm}^3}{1 \text{ L}} = 322 \text{ dm}^3$$

19) $4.44 \text{ lbs} \rightarrow \text{kg}$

$$\frac{4.44 \text{ lbs}}{2.2 \text{ lbs}} \times \frac{453.6 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ kg}}{1000 \text{ g}} = 2.01 \text{ kg}$$

20) $5.55 \times 10^{10} \text{ cm}^3 \rightarrow \text{m}^3$

$$\frac{5.55 \times 10^{10} \text{ cm}^3}{1 \text{ cm}^3} \times \frac{1 \text{ m}^3}{1 \times 10^6 \text{ cm}^3} = 55,500 \text{ m}^3$$

21) $75 \text{ mi/hr} \rightarrow \text{m/s}$

$$\frac{75 \text{ mi}}{1 \text{ hr}} \times \frac{1609 \text{ m}}{1 \text{ mi}} \times \frac{1 \text{ hr}}{3600 \text{ s}} = 34 \text{ m/s}$$

22) $6.7 \text{ lb/ft}^3 \rightarrow \text{g/mL}$

$$\frac{6.7 \text{ lb}}{\text{ft}^3} \times \frac{453.6 \text{ g}}{1 \text{ lb}} \times \frac{1 \text{ ft}^3}{1728 \text{ in}^3} \times \frac{1 \text{ in}^3}{16.39 \text{ cm}^3} \times \frac{1 \text{ mL}}{1 \text{ cm}^3}$$