

## **Ch. 10: Photosynthesis: Energy from the Sun**

1. Identifying Photosynthetic Reactants and Products
2. The Two Pathways of Photosynthesis: An Overview
3. The Interactions of Light and Pigments
4. The Light Reactions: Electron Transport, Reductions, and Photophosphorylation
5. Making Carbohydrate from CO<sub>2</sub>: The Calvin–Benson Cycle
6. Metabolic Pathways in Plants

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## **Ch.10 Objectives:**

### **❖ Students should be able to .....**

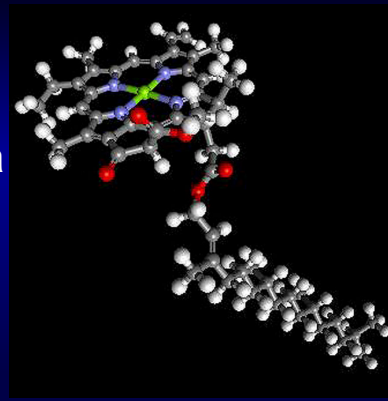
1. **Ch. 10:** Diagram and explain how light-harvesting pigments capture light, and how this is **converted to cellular energy** by noncyclic photosynthesis.
2. \*\*\* Compare & contrast the **energy conversion mechanisms of cellular respiration** and the **light reactions** of photosynthesis. What energy inputs & outputs are used, and in what forms?
3. Describe and diagram the relationship between the **light reactions** and the **“dark” reactions** of photosynthesis.

**❖ Objectives and Study Guide Questions are your HOMEWORK between classes!! Due at the end of the week during Lecture!!**

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## 10.1) Identifying Photosynthetic Reactants and Products

- **Photosynthesis** = the biochemical process by which plants/algae/bacteria capture energy from sunlight and store it in carbohydrates.

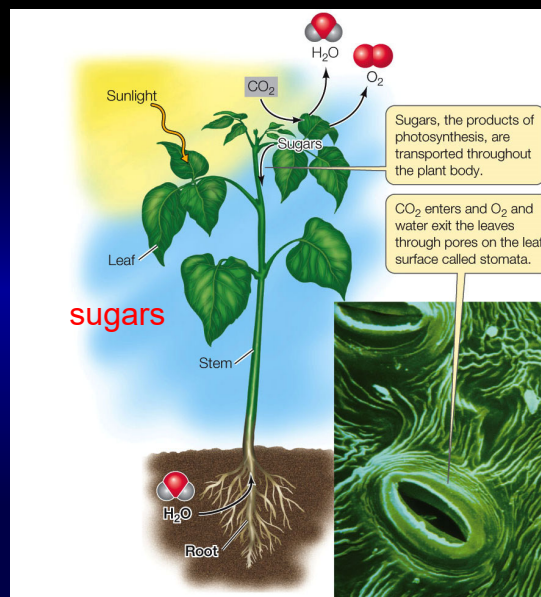


– = *the very basis of life on Earth!!*

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### Figure 10.1 The Ingredients for Photosynthesis

- By the 1800s, scientists had learned:
  1. Three ingredients are needed for photosynthesis: **water, CO<sub>2</sub>, & light.**
  2. There are two products: **carbohydrates & O<sub>2</sub>.**
  3. The water, which comes primarily from the soil, is transported through the **roots** to the **leaves.**
  4. The CO<sub>2</sub> is taken in (and O<sub>2</sub> exits) from the air through **stomata**, or pores, in the leaves.

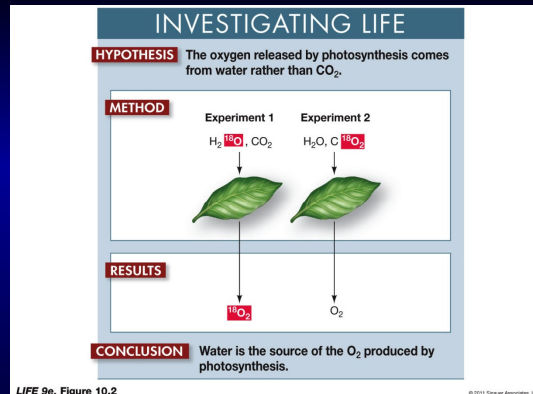


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## Photosynthetic Reactants & Products

- By 1804, scientists had summarized the overall chemical reaction of photosynthesis:
  - ❖  $\text{CO}_2 + \text{H}_2\text{O} + \text{light energy} \rightarrow \text{sugar} + \text{O}_2$



- More recently, using  $\text{H}_2\text{O}$  and  $\text{CO}_2$  labeled with radioactive isotopes, we know that the actual reaction is:



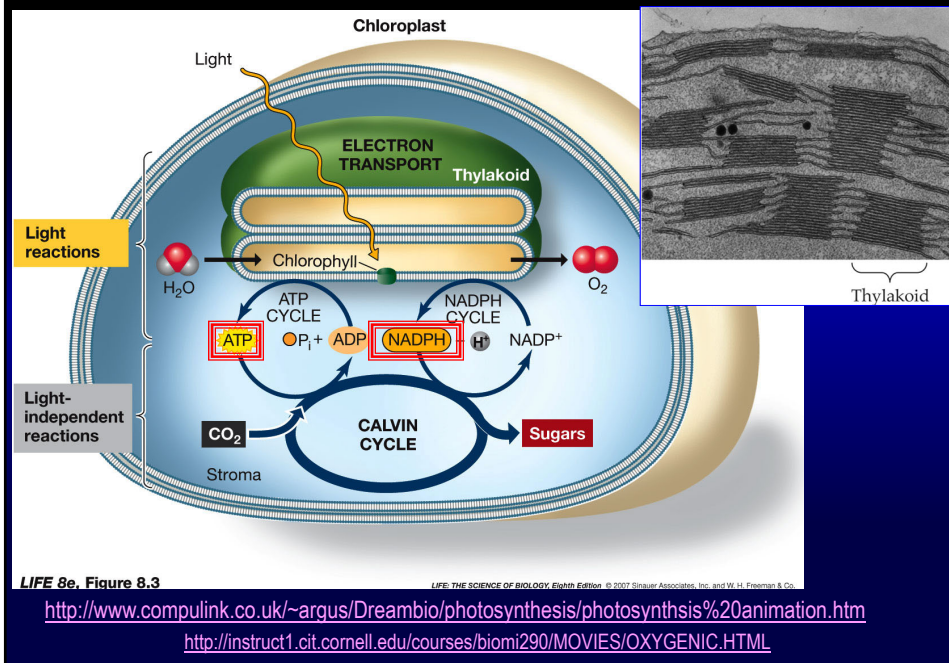
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## 10.2) The Two Pathways of Photosynthesis: An Overview

- A. Photosynthesis** occurs in the **chloroplasts**
- green plant cells, algae (**photoautotrophs**)
  - photoautotrophic bacteria
  - consists of many reactions.
- B. Photosynthesis** can be divided into two pathways:
- The **light reaction** is driven by light energy captured by chlorophyll.
    - It produces **ATP** and **NADPH + H<sup>+</sup>**.
  - The **Calvin–Benson cycle (light-independent rxns/ dark rxns)** does not use light directly.
    - It uses **ATP**, **NADPH + H<sup>+</sup>**, and **CO<sub>2</sub>** to produce sugars.

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## Figure 10.3 An Overview of Photosynthesis



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## 10.3) The Interactions of Light & Pigments

- ❖ Visible light is part of the electromagnetic radiation spectrum.
  - comes in discrete packets called **photons**.
  - also behaves as if it were a **wave**.
- ❖ Two things are required for photons to be active in a biological process:
  1. Photons **must be absorbed** by receptive molecules.
  2. Photons must have **sufficient energy** to perform the chemical work required.

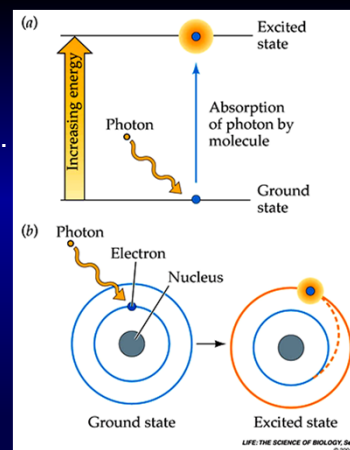
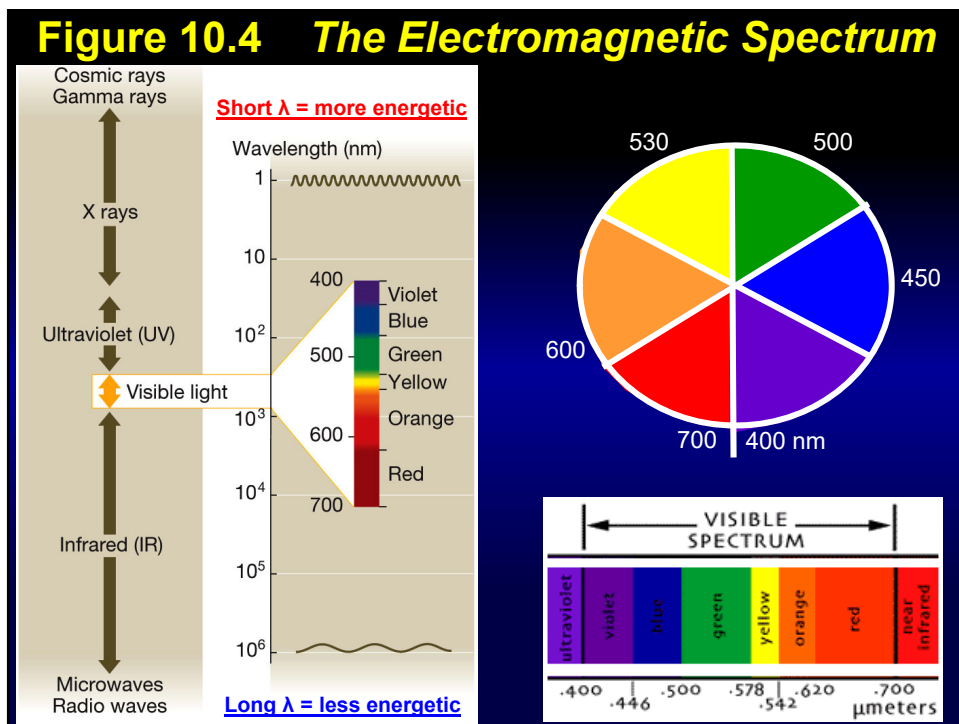


Figure 10.5 Exciting a Molecule

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## A. Light Activation of a Molecule

- When a photon and a pigment molecule meet, one of three things happens:
  1. The photon may **bounce off**,
  2. **pass through**, or
  3. **be absorbed** by the molecule.
- If **absorbed**, the energy of the photon is acquired by the molecule.
- The molecule is **raised from its ground state to an excited state of higher energy**.

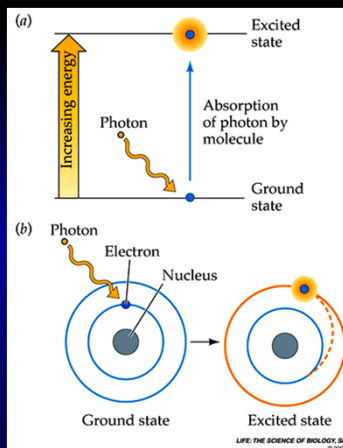


Figure 10.5 Exciting a Molecule

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## B. Pigment Spectra

❖ *A molecule can absorb radiant energy of only certain wavelengths.*

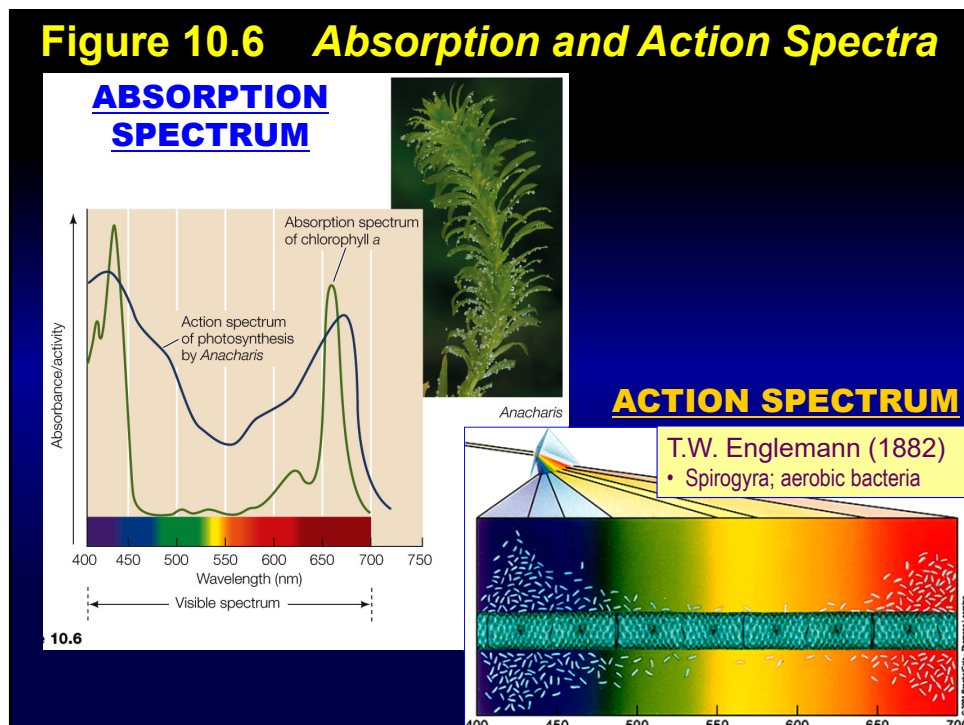
1. If we plot the absorption by the compound as a function of wavelength, the result is an Absorption Spectrum.

2. If absorption results in a biological activity:

a) then a plot of the effectiveness of the light as a function of wavelength

b) is called an Action Spectrum.

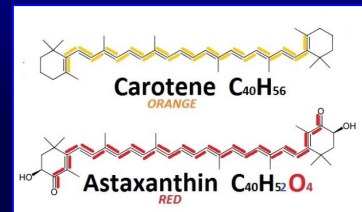
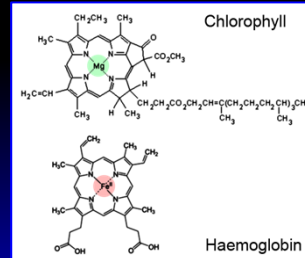
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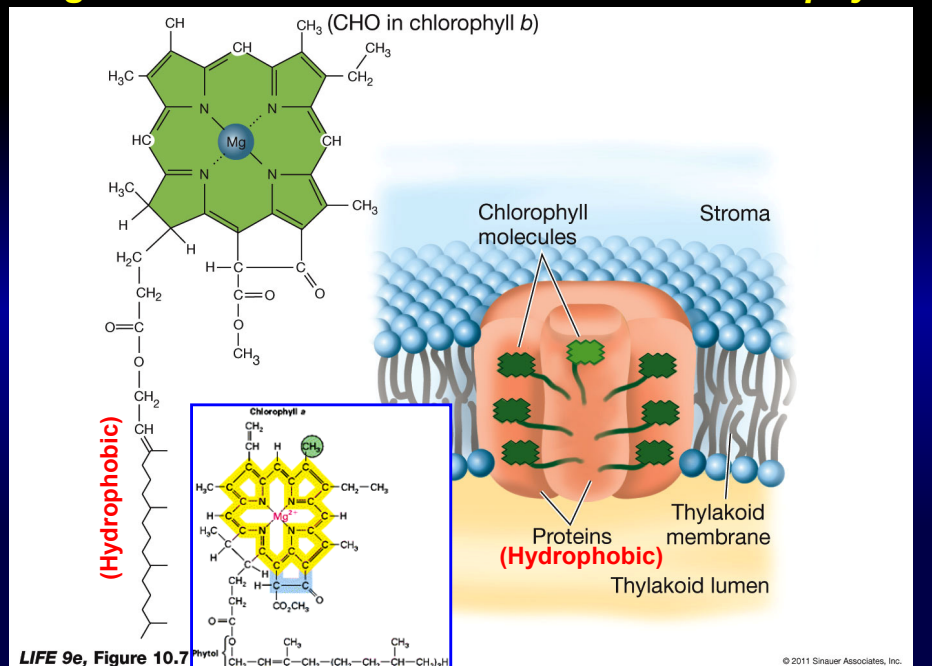
# C. Photosynthetic Pigments

- Plants have two predominant chlorophylls: **chlorophyll a** and **chlorophyll b**.
  - absorb Blue and Red wavelengths
    - Are near the ends of the visible spectrum.
- Other **accessory pigments** absorb photons between the red and blue wavelengths
  - transfer a portion of that energy to chlorophylls.
  - Eg: the **carotenoids**, such as  **$\beta$ -carotene**.



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Figure 10.7 Molecular Structure of Chlorophyll



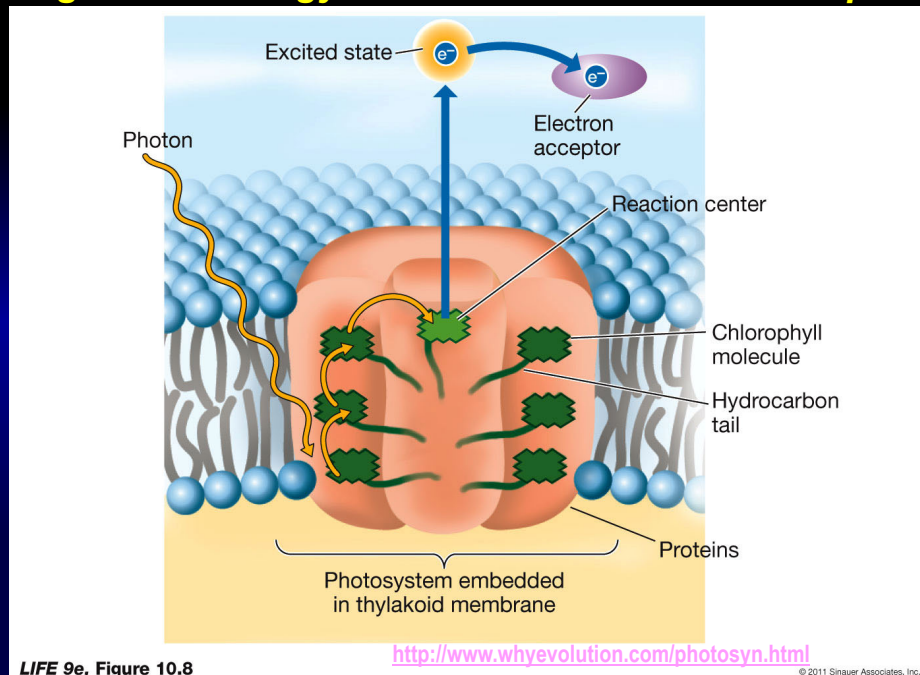
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## D. Interactions of Light and Pigments

- When it absorbs a photon, a pigment molecule enters an unstable **excited state**.
  - molecule may return to the **ground state**.
    - If returns to ground:
      - some of the absorbed energy is given off as **heat**.
    - the rest is **given off as light energy**, or **fluorescence**.
    - If fluorescence does not occur:
      - pigment molecule may pass some absorbed energy to **other pigment molecules**.
- Pigments in photosynthetic organisms are arranged into **antenna systems** ("**light harvesting complexes**").
  - pigments are **packed together & attached to thylakoid membrane proteins**
  - to enable the transfer of energy.
- The excitation energy is passed to the **Reaction Center** of the antenna complex.
  - In plants, the pigment molecule in the center is always a molecule of **chlorophyll a**.

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**Fig. 10.8 Energy Transfer and Electron Transport**

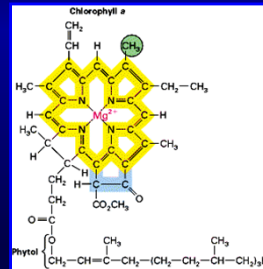


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# Light and Pigments

- **Excited chlorophyll (Chl\*)** in the reaction center acts as a **reducing agent**.
  - electrons of an excited molecule are less tightly held by the nucleus (less stable!!),
  - and more likely to be passed on in a redox reaction to an oxidizing agent.
- Chl\* can react with an oxidizing agent in a reaction such as:
  - $\text{Chl}^* + \text{A} \rightarrow \text{Chl}^+ + \text{A}^-$
- **Chlorophyll becomes a reducing agent** in a redox reaction.



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## 10.4) The Light Reactions: Electron Transport, Reductions, & Photophosphorylation

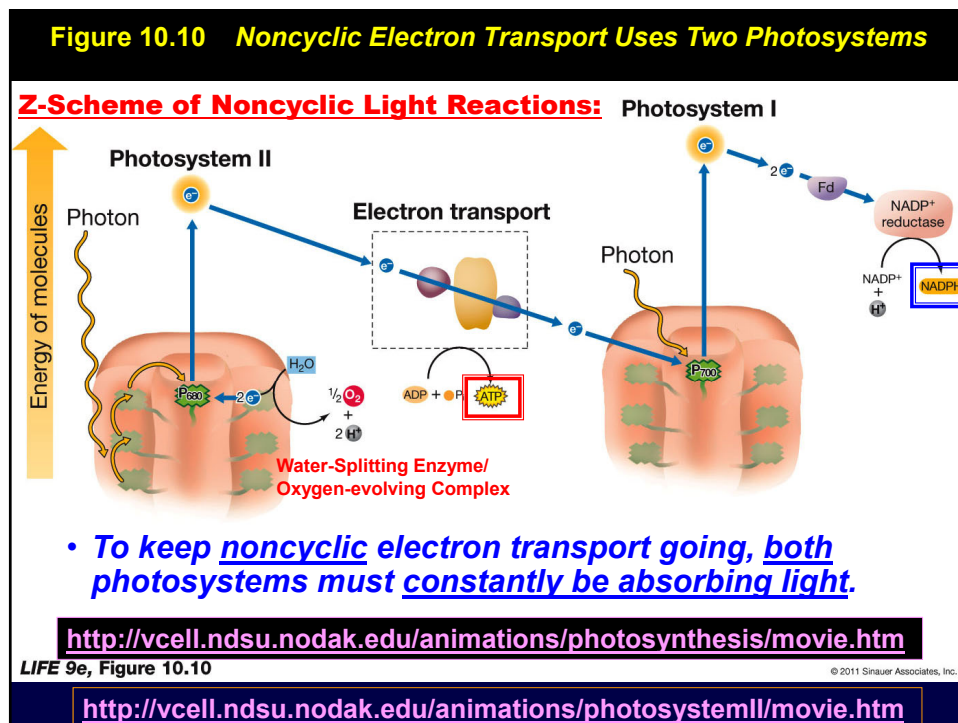
1. The energized electron that leaves the Chl\* in the reaction center immediately participates in a series of redox reactions.
  - The electron flows through a series of carriers in the **thylakoid membrane**, = electron transport.
  - 2 energy rich products of the light reactions result:
    - NADPH + H<sup>+</sup> and ATP, are the result.
2. Chemiosmotic synthesis of ATP in the thylakoid membrane is called **PhotoPhosphorylation**.
3. There are two different systems for transport of electrons in photosynthesis.
  - a) **Noncyclic Electron Transport** produces **NADPH** + H<sup>+</sup> and **ATP** & **O<sub>2</sub>**.
  - b) **Cyclic Electron Transport** produces **only ATP**.

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## A. Light Reactions: Noncyclic Electron Transport

- ❖ In **NONCYCLIC Electron Transport**, two photosystems are needed.
- ❖ **Photosystems** = light-driven molecular units.
  - consist of **many chlorophyll molecules and accessory pigments**.
  - **bound to proteins** in separate energy-absorbing antenna systems.
- 1. **Photosystem I** uses light energy to reduce  $\text{NADP}^+ \rightarrow \text{NADPH} + \text{H}^+$ .
  - The **reaction center** contains a **chlorophyll a** molecule called **P<sub>700</sub>**
    - it best absorbs light at a wavelength of **700 nm**.
- 2. **Photosystem II** uses light energy to oxidize water.
  - producing electrons, protons, and O<sub>2</sub>.
  - The **reaction center** contains a **chlorophyll a** molecule called **P<sub>680</sub>**
    - best absorbs light at a wavelength of **680 nm**.

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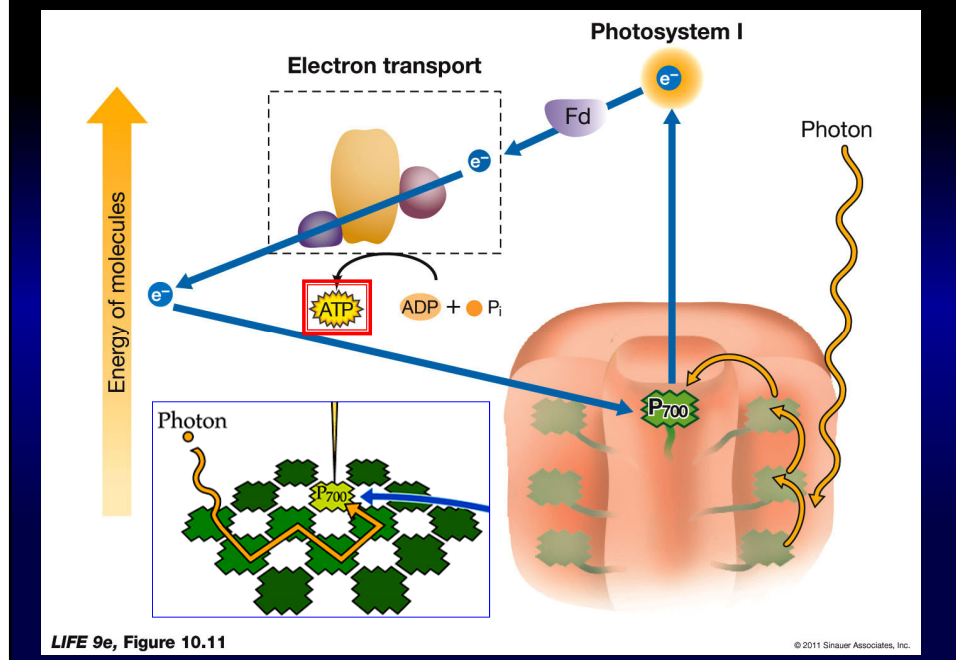
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## B. Light Reactions: Cyclic Electron Transport

- Cyclic** electron transport produces only ATP.
  - called cyclic, because:
  - the electron passed from an excited  $P_{700}$  molecule cycles back to the same  $P_{700}$  molecule.
- Water does not enter into the cyclic electron flow reactions, and no  $O_2$  is released.**
- In cyclic electron flow, **Photosystem-I acts on its own**.

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**Fig. 10.11 CYCLIC Electron Transport Traps Light Energy as ATP**



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## C. Light Reactions: Photophosphorylation

- ATP is produced by a Chemiosmotic Mechanism similar to that of mitochondria, called PhotoPhosphorylation.
  - High-energy electrons move through a series of redox reactions
  - release energy used to transport protons across the membrane.
- Active proton transport results in the Proton-Motive Force: a difference in pH and electric charge across the membrane.

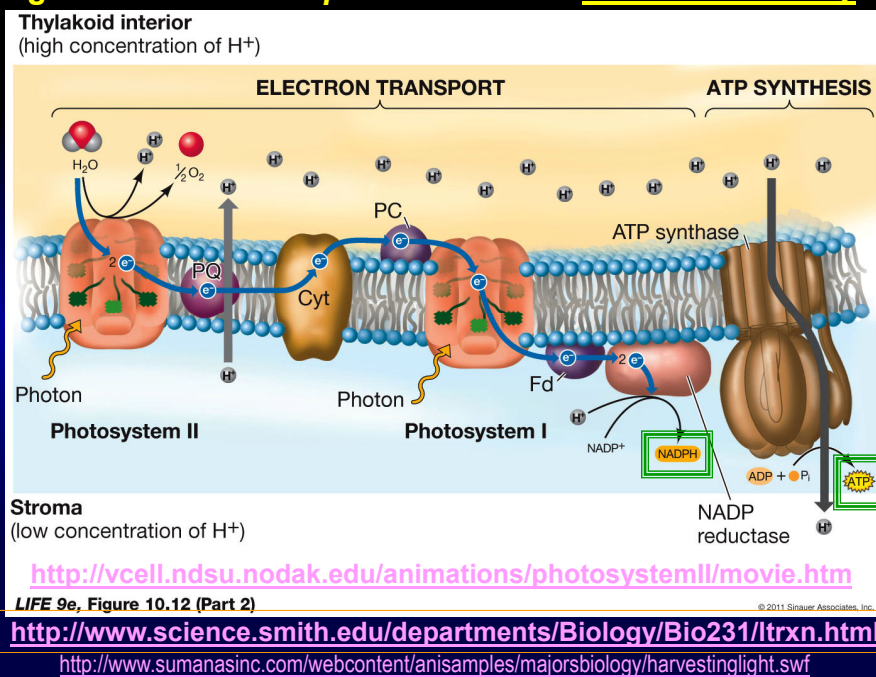
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## Light Reactions: Photophosphorylation & Chemiosmosis

- The electron carriers in the thylakoids **move protons into the interior of the thylakoid (lumen!)**
  - the **inside** becomes **acidic** with respect to the outside.
- This difference in pH:
  - leads to the diffusion of H<sup>+</sup> out of the thylakoid
  - through specific protein channels, **ATP Synthases**.
- The **ATP Synthases couple** the formation of ATP to proton diffusion back across the thylakoid membrane.
  - \*\* JUST LIKE Ox. Phos'n in Mitochondria!!!! \*\***

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### Figure 10.12 Chloroplasts Form ATP Chemiosmotically



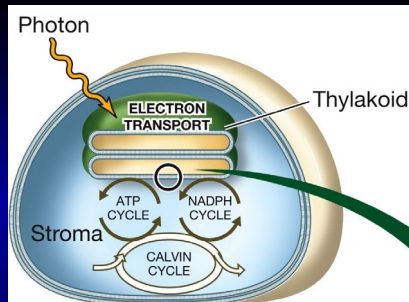
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## 10.5) Making Carbohydrate from CO<sub>2</sub>: The Calvin–Benson Cycle

- Calvin-Benson cycle reactions = in stroma.

- does not use sunlight directly;
- but *it requires the ATP and NADPH + H<sup>+</sup> produced in the light reactions*
- these cannot be “stockpiled”.

- *Calvin-Benson reactions require light indirectly & take place only in the presence of light!* (not really “Dark” rxns!)



Energy Transfer Molecule Type:	Mostly for Catabolism	Mostly for Anabolism
Triphosphate Nucleotides	ATP	GTP
High-Energy Electron Carriers	NADH (and FADH <sub>2</sub> )	NADPH

<http://www.science.smith.edu/departments/Biology/Bio231/ltrxn.html>

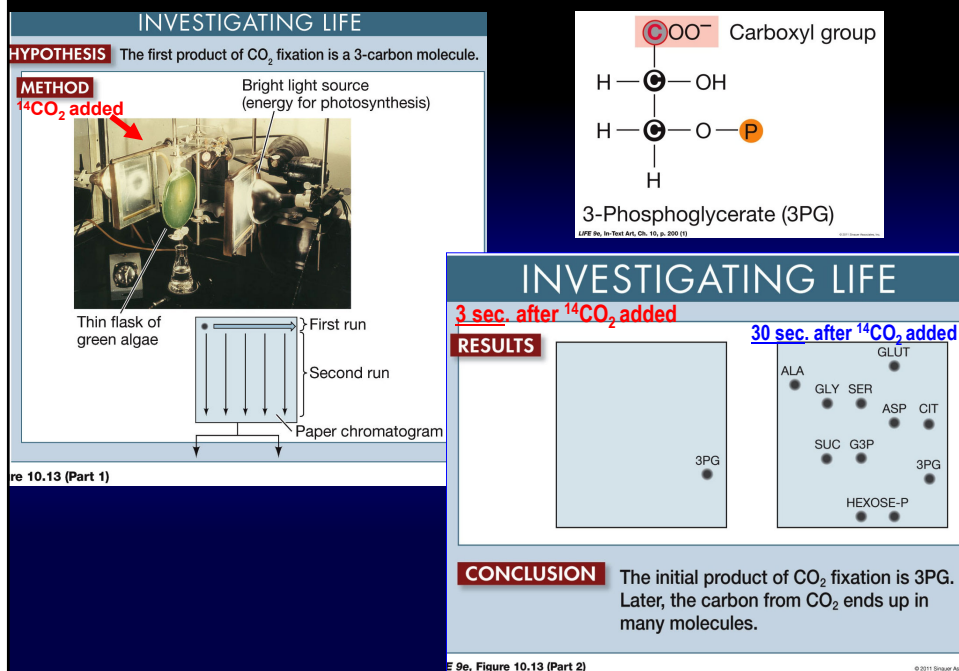
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## A. The Calvin–Benson Cycle: Experimental Discovery

- Experiments showed:
  - steps of the Calvin–Benson cycle required **radioactively labeled carbon in  $^{14}\text{CO}_2$** .
- Exposure of ***Chlorella*** (unicellular algae) cells to  $^{14}\text{CO}_2$  for **3 seconds**:
  - one compound was labeled with  $^{14}\text{C}$ .
  - a 3-C sugar called **3-phosphoglycerate (3PG)**.
- Other products of the cycle:
  - found by stepwise increases in time of exposure
  - until the whole pathway was revealed.

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### Figure 10.13 Tracing the Pathway of $\text{CO}_2$



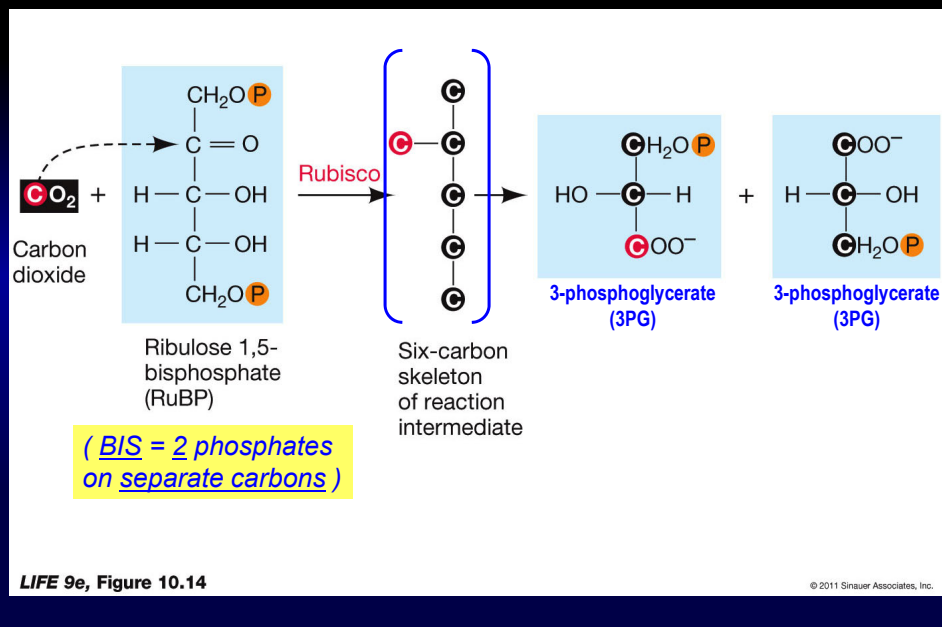
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## B. Fixation of CO<sub>2</sub>: The Calvin–Benson Cycle

1. The initial reaction of the Calvin–Benson cycle **fixes one CO<sub>2</sub> into a 5-carbon compound, ribulose 1,5-bisphosphate (RuBP).**
2. An unstable **intermediate 6-carbon compound** forms → breaks down into **two 3-C molecules of 3PG.**
3. The enzyme that catalyzes the fixation of CO<sub>2</sub> is **ribulose biphosphate carboxylase/oxygenase**, called **RUBISCO**. (**Carbon-Fixation!!**)
4. **Rubisco is the most abundant protein in the world!!**

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**Figure 10.14 RuBP Is the Carbon Dioxide Acceptor**



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## Fixation of CO<sub>2</sub>: The Calvin–Benson Cycle

- The Calvin–Benson cycle consists of three processes:
  1. **Fixation of CO<sub>2</sub>**, by combination with **RuBP** (catalyzed by **RUBISCO**).
  2. Conversion of fixed CO<sub>2</sub> into carbohydrate (3PG) (*this step uses ATP and NADPH*).
  3. Regeneration of the CO<sub>2</sub> acceptor RuBP (5/6 of the carbon fixed!!!) by ATP.

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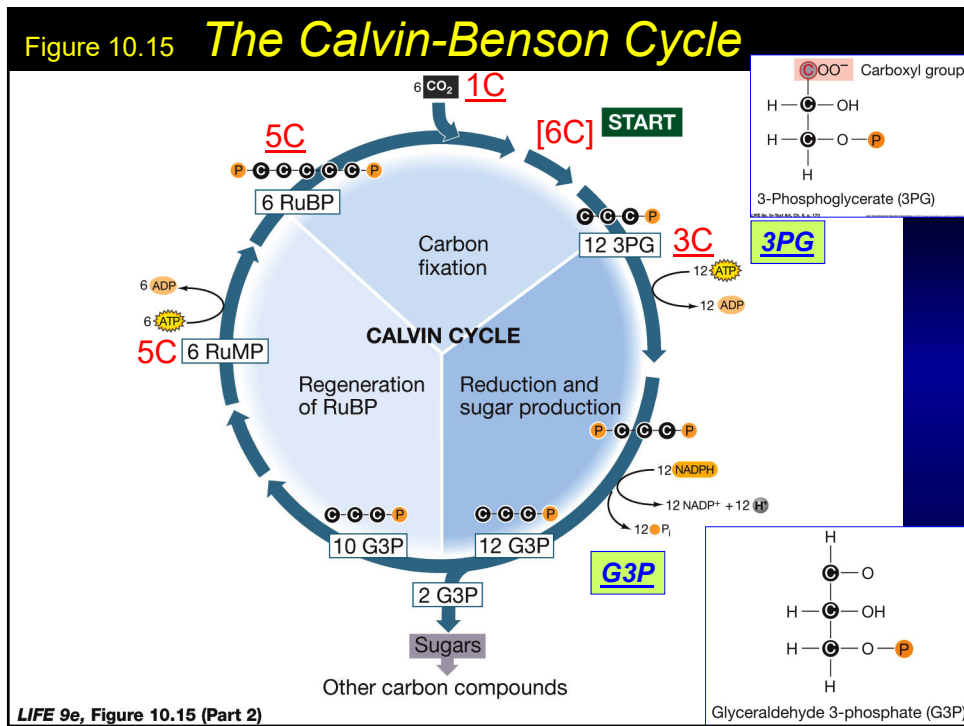
## C. The Calvin–Benson Cycle: Products

- The end product of the cycle is **glyceraldehyde 3-phosphate, G3P**.
- There are two fates for the stored G3P:
  - **1/3** ends up as **starch**, which is stored in the chloroplast and serves as a source of glucose.
  - **2/3** is converted to the disaccharide **sucrose**, which is transported to other organs.

<http://www.science.smith.edu/departments/Biology/Bio231/calvin.html>

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## Making Carbohydrate from $\text{CO}_2$ : The Calvin–Benson Cycle

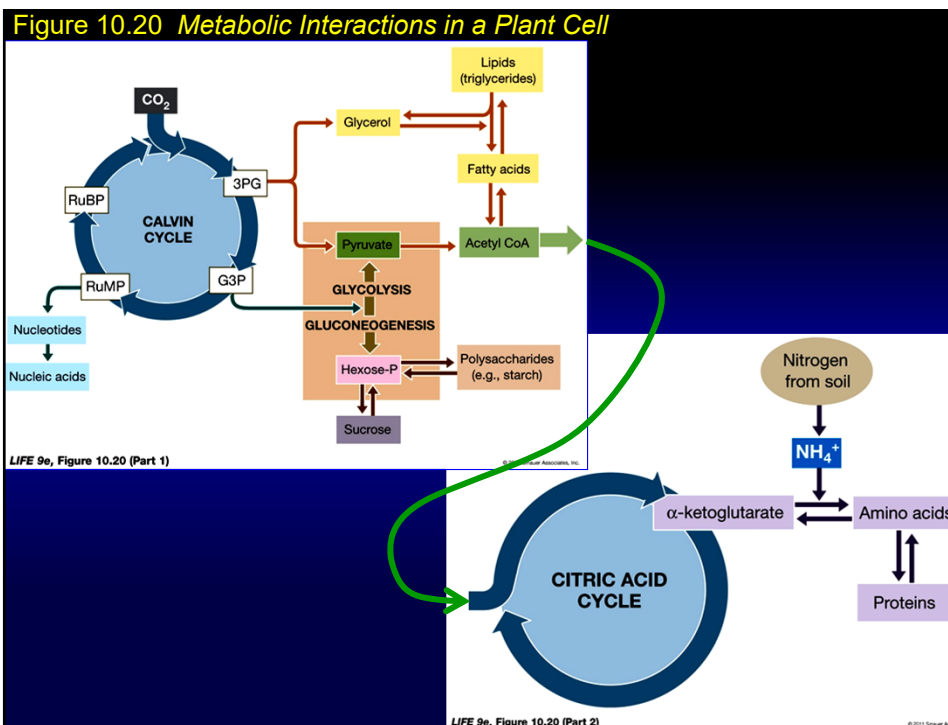
- Products of the Calvin–Benson cycle:
  - are vitally important to the **biosphere**:
  - they are the total energy yield from sunlight conversion by green plants.
- ❖ **Most of the stored energy is released by the plant's own Glycolysis & Cellular Respiration.**
  - Some of the carbon of glucose becomes **part of amino acids, lipids, & nucleic acids**.
  - Some of the stored energy is **consumed by heterotrophs**, where glycolysis and respiration release the stored energy.

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## 10.6) Metabolic Pathways in Plants

- Green plants are **autotrophs** = “Producers”!!
  - synthesize all their molecules from 3 simple starting materials: **CO<sub>2</sub>, H<sub>2</sub>O, & NH<sub>4</sub>**.
- To satisfy their need for ATP, plants, like all other organisms, carry out **cellular respiration**.
  - Both aerobic respiration and fermentation can occur in plants
  - respiration is more common.
- Cellular respiration takes place **both in dark and in light**.
- Photosynthesis and respiration = closely linked by the Calvin–Benson cycle.
  - Some **G3P** from Calvin–Benson → **pyruvate**, the end product of glycolysis.
- Some G3P → **hexose phosphates**, which can enter glycolysis.
  - Once **carbon skeletons** from the Calvin–Benson cycle enter **glycolysis** and the **TCA cycle**, they can be **used to make lipids, proteins, and other carbohydrates**.

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# Metabolic Pathways in Plants

1. Energy flows from **sunlight** to **reduced carbon** in photosynthesis to **ATP** in respiration.
2. Energy can be stored in **macromolecules** such as polysaccharides, lipids, and proteins.
3. For plants *to grow*,
  - a) **energy storage must exceed energy released** or
  - b) overall **carbon fixation** by photosynthesis must exceed respiration.
4. The capture and movement of sun energy becomes the **basis for ecological food chains**.

